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## Crystal Structure

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# A zigzag chain structure in catena-poly[[[tetra- $\mu$-acetamidato- $\kappa^{4} N$ : $O$;$\kappa^{4} O: N$-dirhodium(II,III)]- $\mu$-chloro] methanol solvate] 

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In $\left\{\left[\mathrm{Rh}_{2}\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{NO}\right)_{4} \mathrm{Cl}\right] \cdot \mathrm{CH}_{3} \mathrm{OH}\right\}_{n}$, the cationic dirhodium complex and bridging chloro ligands form a one-dimensional zigzag chain, $\left[\mathrm{Rh}_{2}(\text { acam })_{4}(\mu-\mathrm{Cl})\right]$ (Hacam is acetamide). There is a large difference between the two $\mathrm{Rh}-\mathrm{Cl}$ distances [2.6076 (14) and $2.5027(14) \AA$ ). Neighboring chains are connected by two $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds per link between the amidate ligands.

## Comment

A halide-bridged one-dimensional chain is commonly observed in structures with metal-metal-bonded paddlewheel complexes, such as $\mathrm{K}\left[\mathrm{Mo}_{2}\left(\mathrm{O}_{2} \mathrm{CH}\right)_{4} \mathrm{Cl}\right]$ (Robbins \& Martin, 1984), $\left[\mathrm{Pt}_{2}\left(\mathrm{~S}_{2} \mathrm{CR}\right)_{4} \mathrm{I}\right]$ (Bellitto et al., 1983; Kitagawa et al., 2001; Mitsumi et al., 2002) and $\left[\mathrm{Ru}_{2}\left(\mathrm{O}_{2} \mathrm{CR}\right)_{4} X\right](X=\mathrm{Cl}$ and Br ; Angaridis, 2005). The chain in $\mathrm{K}\left[\mathrm{Mo}_{2}\left(\mathrm{O}_{2} \mathrm{CH}\right)_{4} \mathrm{Cl}\right]$ is a zigzag one, that in $\left[\mathrm{Pt}_{2}\left(\mathrm{~S}_{2} \mathrm{CR}\right)_{4} \mathrm{I}\right]$ is linear and $\left[\mathrm{Ru}_{2}\left(\mathrm{O}_{2} \mathrm{CR}\right)_{4} X\right]$ contains both types. We have reported assemblies of acet-amidate-bridged dirhodium paddlewheel complexes with halide linkers, viz. one-dimensional chain structures in $\left\{\left[\mathrm{Rh}_{2}(\operatorname{acam})_{4}(\mu-X)\right] \cdot \mathrm{H}_{2} \mathrm{O}\right\}_{n}$ (Hacam is acetamide; $X=\mathrm{Cl}, \mathrm{Br}$ and I; $n=0,2,3$ and 7; Yang et al., 2000, 2001), a twodimensional honeycomb structure in $\left[\left\{\mathrm{Rh}_{2}(\mathrm{acam})_{4}\right\}_{3}\left(\mu_{3^{-}}\right.\right.$ $\left.\mathrm{Cl})_{2}\right] \cdot 4 \mathrm{H}_{2} \mathrm{O}$ (Takazaki et al., 2003) and a three-dimensional diamondoid structure in $\left[\left\{\mathrm{Rh}_{2}(\text { acam })_{4}\right\}_{2}\left(\mu_{4}-\mathrm{I}\right)\right] \cdot 6 \mathrm{H}_{2} \mathrm{O}$ (Fuma et al., 2004). In the zigzag chain structure of $\left\{\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}(\mu\right.\right.$ $\left.X)] \cdot \mathrm{H}_{2} \mathrm{O}\right\}_{n}$, the $\mathrm{Rh}-\mathrm{Cl}-\mathrm{Rh}$ angle varies with the hydrogen bonding involving the water molecules. We have attempted to synthesize a chain structure with solvent molecules other than water. In this paper, we report the zigzag chain structure of $\left\{\left[\mathrm{Rh}_{2}(\text { acam })_{4}(\mu-\mathrm{Cl})\right] \cdot \mathrm{CH}_{3} \mathrm{OH}\right\}_{n}$, (I).

The structure of (I) is shown in Fig. 1. There are one independent $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4} \mathrm{Cl}\right]$ unit and one methanol molecule in the asymmetric unit. The bond distances in the $\left[\mathrm{Rh}_{2^{-}}\right.$ (acam) ${ }_{4}$ ] skeleton are similar to those observed previously for
the cationic $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}\right]$ unit (Yang et al., 2000, 2001, 2006; Ebihara \& Fuma, 2006; Baranovskii et al., 1986). The $\left[\mathrm{Rh}_{2}(\text { acam })_{4} \mathrm{Cl}\right]$ unit forms an infinite zigzag chain structure (Fig. 2) as in $\left[\mathrm{Rh}_{2}(\operatorname{acam})_{4}(\mu-\mathrm{Cl})\right]$ and $\left[\mathrm{Rh}_{2}(\operatorname{acam})_{4}(\mu-\mathrm{Cl})\right]$ ]$7 \mathrm{H}_{2} \mathrm{O}$ (Yang et al., 2000, 2001).


In the reported zigzag chain structures of $\left[M_{2} L_{4} X\right][M=\mathrm{Ru}$ and $\mathrm{Rh}, L=\mathrm{O}_{2} \mathrm{C} R$ and $\mathrm{HN}(\mathrm{O}) \mathrm{C} R$, and $X=\mathrm{Cl}, \mathrm{Br}$ and I ; Bennett et al., 1969; Togano et al., 1980; Kimura et al., 1982; Chakravarty \& Cotton, 1985; Chakravarty et al., 1985; Das \& Chakravarty, 1991; Abe et al., 1992; Barral et al., 1998, 1999, 2000, 2004; Cukiernik et al., 1998; Yang et al., 2000, 2001], the dimetal unit usually lies on an inversion center or on a twofold axis. For example, in the chain structure of $\left[\mathrm{Rh}_{2}(\text { acam })_{4}(\mu\right.$ $\mathrm{Cl})$ ] (Yang et al., 2001), which crystallizes in $C 2 / c$, the dirhodium unit lies on an inversion center and the Cl atom lies on a twofold axis. In the structure of $\left[\mathrm{Ru}_{2}\left(\mathrm{O}_{2} \mathrm{CC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right)_{4} \mathrm{Cl}\right]$ (Das \& Chakravarty, 1991), there are three independent diruthenium units, of which two lie on inversion centers and one occupies a general position. In the structure of (I), the chain is propagated along the $c$ axis with each unit of the complex connected to adjacent units generated by the $c$-glide plane at $y=\frac{1}{4}$. Compound (I) is the first example of a zigzag chain


Figure 1
The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level and H atoms are shown as small spheres of arbitrary radii. [Symmetry code: (i) $x$, $-y+\frac{1}{2},-\frac{1}{2}+z$.]
constructed from one independent paddlewheel complex that does not have any symmetry.

The $\mathrm{Rh}-\mathrm{Cl}-\mathrm{Rh}$ angles in $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}(\mu-\mathrm{Cl})\right]$ and $\left[\mathrm{Rh}_{2^{-}}\right.$ (acam) $\left.)_{4}(\mu-\mathrm{Cl})\right] \cdot 7 \mathrm{H}_{2} \mathrm{O}$ are $115.48(10)$ and $153.50(6)^{\circ}$, respectively. Since the corresponding angle in (I) is $114.59(5)^{\circ}$, the chain structure in (I) more closely resembles that in $\left[\mathrm{Rh}_{2}(\text { acam })_{4}(\mu-\mathrm{Cl})\right]$. Hydrogen bonds from the NH groups of the acam ligands to the O atoms of the next complexes along the chain $\left[\mathrm{N} 1 \cdots \mathrm{O} 4^{\mathrm{i}}\right.$ and $\mathrm{N} 3 \cdots \mathrm{O} 2^{\mathrm{ii}}$; symmetry codes: (i) $x$, $-y+\frac{1}{2}, z-\frac{1}{2}$; (ii) $x,-y+\frac{1}{2}, z+\frac{1}{2}$; Table 2] also support the chain structure, as was observed in $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}(\mu-\mathrm{Cl})\right]$ (Yang et al., 2001). The methanol molecule lies beside the chain, accepting a hydrogen bond from an N atom of an acam ligand ( N 2 ) and donating a hydrogen bond to an O atom of a neighboring complex ( $\mathrm{O} 2^{\mathrm{ii}}$ ).

The $\mathrm{Rh}-\mathrm{Rh}-\mathrm{Cl}$ angles in (I) (Table 1) are slightly more bent than that in $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}(\mu-\mathrm{Cl})\right]\left[174.52(4)^{\circ}\right]$. The $\mathrm{Rh} 1-$ Cl 1 and $\mathrm{Rh} 2-\mathrm{Cl} 1^{\mathrm{i}}$ distances (Table 1) are both different from the corresponding value in $\left[\mathrm{Rh}_{2}(\mathrm{acam})_{4}(\mu-\mathrm{Cl})\right][2.581(1) \AA]$. The difference between these $\mathrm{Rh}-\mathrm{Cl}$ distances is very large (ca $0.1 \AA$ ). In the previously reported chain structures, the largest differences (ca $0.04 \AA$ ) between $M-\mathrm{Cl}$ bonds were observed in $\left[\mathrm{Ru}_{2}\left\{\mathrm{HN}(\mathrm{O}) \mathrm{CC}_{6} \mathrm{H}_{4} R\right\}_{4}(\mu-\mathrm{Cl})\right][R=\mathrm{H}$ (Chakravarty \& Cotton, 1985) and $R=\mathrm{Cl}$ (Chakravarty et al., 1985)]. The long-short pattern of the $M-\mathrm{Cl}$ bonds in the $\mathrm{Cl}-M M-$ $\mathrm{Cl}-M M-\mathrm{Cl}$ unit is long-long-short-short for $\left[\mathrm{Ru}_{2}\{\mathrm{HN}(\mathrm{O})\right.$ $\left.\left.\mathrm{CC}_{6} \mathrm{H}_{4} R\right\}_{4}(\mu-\mathrm{Cl})\right]$, since two independent diruthenium units lie on inversion centers, but long-short-long-short for (I).

The chains are mutually parallel and are connected to each other by a pair of hydrogen bonds [ $\mathrm{N} 4 \cdots \mathrm{O} 3^{\mathrm{iii}}$ and $\mathrm{N} 4{ }^{\text {iiii }} \ldots \mathrm{O} 3$; symmetry code: (iii) $-x+1,-y+1,-z+1]$. These interchain hydrogen bonds were not observed in other chain structures with amidate-bridged paddlewheel complexes (Chakravarty \& Cotton, 1985; Chakravarty et al., 1985; Yang et al., 2000, 2001).


Figure 2
The crystal structure of (I). Hydrogen bonds are drawn as thin lines. [Symmetry codes: (i) $x,-y+\frac{1}{2}, z-\frac{1}{2}$; (ii) $x,-y+\frac{1}{2}, z+\frac{1}{2}$; (iii) $-x+1,-y+1$, $-z+1$.]

## Experimental

$\left[\mathrm{Rh}_{2}(\text { acam })_{4}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right] \mathrm{ClO}_{4}$ was prepared according to the method described by Baranovskii et al. (1986). Into a methanol solution of $\left[\mathrm{Rh}_{2}(\text { acam })_{4}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right] \mathrm{ClO}_{4}\left(3.1 \mathrm{mmol} \mathrm{l}^{-1}\right)$, a methanol solution of $\mathrm{CoCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}\left(0.21 \mathrm{~mol} \mathrm{l}^{-1}\right)$ was diffused slowly. After several days, brown crystals of (I) were obtained.

Crystal data
$\left[\mathrm{Rh}_{2}\left(\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{NO}\right)_{4} \mathrm{Cl}\right] \cdot \mathrm{CH}_{4} \mathrm{O}$
$M_{r}=505.56$
Monoclinic, $P 2_{1} / c$
$a=8.601$ (4) $\AA$
$b=14.254$ (7) $\AA$
$c=12.664$ (7) $\AA$
$\beta=98.854$ (5) ${ }^{\circ}$
$V=1534.1(13) \AA^{3}$

## Data collection

Rigaku/MSC Mercury CCD diffractometer
$\omega$ scans
Absorption correction: integration (NUMABS; Higashi, 1999) $T_{\text {min }}=0.624, T_{\text {max }}=0.755$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.034$
$w R\left(F^{2}\right)=0.071$
$S=1.17$
3496 reflections
198 parameters
H atoms treated by a mixture of independent and constrained refinement

$$
Z=4
$$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=2.189 \mathrm{Mg} \mathrm{~m}^{-3}
\end{aligned}
$$

Mo $K \alpha$ radiation
$\mu=2.35 \mathrm{~mm}^{-1}$
$T=173$ (2) K
Prism, brown
$0.30 \times 0.10 \times 0.10 \mathrm{~mm}$

12335 measured reflections
3496 independent reflections 3246 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.034$
$\theta_{\text {max }}=27.5^{\circ}$

$$
\begin{aligned}
& w=1 / {\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0256 P)^{2}\right.} \\
&+3.6476 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.000 \\
& \Delta \rho_{\max }=0.95 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.80 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\AA \AA^{\circ}$ ).

| Rh1-Rh2 | $2.4247(11)$ | $\mathrm{Rh} 2-\mathrm{Cl} 1^{\mathrm{i}}$ | $2.5027(14)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{Rh} 1-\mathrm{Cl} 1$ | $2.6076(14)$ | $\mathrm{Rh} 2-\mathrm{O} 2$ | $2.057(3)$ |
| $\mathrm{Rh} 1-\mathrm{O} 1$ | $2.026(3)$ | $\mathrm{Rh} 2-\mathrm{O} 3$ | $2.051(3)$ |
| $\mathrm{Rh} 1-\mathrm{O} 4$ | $2.052(3)$ | $\mathrm{Rh} 2-\mathrm{N} 1$ | $1.978(3)$ |
| $\mathrm{Rh} 1-\mathrm{N} 2$ | $1.975(3)$ | $\mathrm{Rh} 2-\mathrm{N} 4$ | $1.968(3)$ |
| $\mathrm{Rh} 1-\mathrm{N} 3$ | $1.979(3)$ |  |  |
| $\mathrm{Rh} 2-\mathrm{Rh} 1-\mathrm{Cl} 1$ | $170.82(2)$ | $\mathrm{Rh} 2^{\mathrm{ii}}-\mathrm{Cl} 1-\mathrm{Rh} 1$ | $114.59(5)$ |
| $\mathrm{Rh} 1-\mathrm{Rh} 2-\mathrm{Cl} 1^{\mathrm{i}}$ | $173.94(2)$ |  |  |
| $\mathrm{O} 1-\mathrm{Rh} 1-\mathrm{Rh} 2-\mathrm{N} 1$ | $1.08(12)$ | $\mathrm{N} 3-\mathrm{Rh} 1-\mathrm{Rh} 2-\mathrm{O} 3$ | $3.73(12)$ |
| $\mathrm{N} 2-\mathrm{Rh} 1-\mathrm{Rh} 2-\mathrm{O} 2$ | $3.40(12)$ | $\mathrm{O} 4-\mathrm{Rh} 1-\mathrm{Rh} 2-\mathrm{N} 4$ | $5.04(12)$ |
| Symmetry codes: (i) $x,-y+\frac{1}{2}, z-\frac{1}{2} ;$; (ii) $x,-y+\frac{1}{2}, z+\frac{1}{2}$. |  |  |  |

Table 2
Hydrogen-bond geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.88 | 2.30 | $3.180(4)$ | 175 |
| $\mathrm{~N} 2-\mathrm{H} 2 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.88 | 2.28 | $2.976(5)$ | 136 |
| $\mathrm{~N} 3-\mathrm{H} 3 \cdots \mathrm{O}^{\mathrm{ii}}$ | 0.88 | 2.43 | $3.251(5)$ | 155 |
| $\mathrm{~N} 4-\mathrm{H} 4 \cdots \mathrm{O}^{3 i}$ | 0.88 | 2.45 | $3.277(4)$ | 158 |
| $\mathrm{O}^{\mathrm{H}}-\mathrm{H} 17 \cdots \mathrm{O}^{\mathrm{ii}}$ | $0.86(6)$ | $2.25(6)$ | $3.045(4)$ | $155(5)$ |

Symmetry codes: (i) $x,-y+\frac{1}{2}, z-\frac{1}{2}$; (ii) $x,-y+\frac{1}{2}, z+\frac{1}{2}$; (iii) $-x+1,-y+1,-z+1$.

The positional parameters of the H atom attached to atom O 5 were refined, with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{O})$. All other H atoms were placed in idealized positions and treated as riding atoms, with $\mathrm{C}-\mathrm{H}$ distances of $0.98 \AA, \mathrm{~N}-\mathrm{H}$ distances of $0.88 \AA$, and $U_{\text {iso }}(\mathrm{H})$ values of $1.5 U_{\text {eq }}(\mathrm{C})$ or $1.2 U_{\text {eq }}(\mathrm{N})$.

## metal-organic compounds

Data collection: CrystalClear (Molecular Structure Corporation \& Rigaku, 2001); cell refinement: CrystalClear; data reduction: TEXSAN (Rigaku/MSC, 2004); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97 and TEXSAN.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FA3040). Services for accessing these data are described at the back of the journal.

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